Younger Granites and Associated Pegmatites of Gabal El Fereyid – Wadi Rahaba Area, South Eastern Desert, Egypt: Geological and Geochemical Characteristics

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Abstract. Lithologically, the rock types of G. El Fereyid - W. Rahaba area are dominantly granites associated with pegmatites. The granitic rocks comprise tonalite and monzogranite with (minor acidic dykes), quartz and pegmatites occur as a vein type (granitic and perthitic pegmatites) hosted within the monzogranite. Structural analysis shows the presence of a prominent set of folds with axes striking WNW to NW and a less prominent set with pronounced ENE and NNW trends. Petrochemical studies and tectonic discrimination diagrams for the monzogranite reveal that it is classified as post orogenic granites (POG). Geochemically, monzogranite shows enrichment (peaks) of large ion lithophile elements (LILE; Pb, Rb, Ba, Sr) and high field strength elements (HFSE; Y, Zr, Nd,) and depletion (troughs) of K, P and Ti. Pegmatite shows enrichment (peaks) of large ion lithophile elements (LILE; Pb, Rb, Sr) and high field strength elements (HFSE; Y, Zr, Nd, elements (LILE; Pb, Rb, Sr) and high field strength elements (HFSE; Y, Zr, Nd, elements (LILE; Pb, Rb, Sr) and high field strength elements (HFSE; Y, Zr, Th, U, Nb) and depletion (troughs) of K, P and Ti. They exhibit high K-calc alkaline magma and peraluminous characteristics and considered as post orogenic granites (POG). The monzogranite reveal small to moderate negative Eu anomaly, while the pegmatites reveal moderate to large negative Eu anomaly. The pegmatite displays clear W-type tetrad effect of the REE while, the studied monzogranite do not show tetrad effect.

Keywords: Petrology, Geochemistry, G. El Fereyid - W. Rahaba granites, Pegmatites, Egypt.

1 Introduction

The investigated area is located in the South Eastern Desert between latitudes $23^{\circ} 15^{\circ} 00^{"}-23^{\circ} 30^{\circ} 00^{"}$ N and longitudes $35^{\circ} 15^{\circ} 00^{"}-35^{\circ} 30^{\circ} 00^{"}$ E. The FCC image of band ratios which covers a major part of study area (Fig.1) is used in regional interpretation for general rock units identification and association. Topographically the area is generally characterized by low to moderate relief, but granitic hills form high peaks of relatively high elevation compared to the surrounding other rock units. Most of the radioactive occurrences in the basement rocks of Egypt are present in the younger granites and associated pegmatites [1]. Many promising criteria including rock compositions, mineralogy, isotopic characteristics and spatial distribution indicate a genetic relationship between granite and the intruding pegmatites [2-4]. These pegmatites have been considered as a favorable uranium and thorium mineralization environments [5].

The high level of radioactivity of these rocks is attributed to the presence of accessory minerals like zircon, monazite, thorite, uranothorite and allanite [6]. The area was studied geologically, structurally, mineralogically, geochemically and radiometrically by numerous authors e.g. [7-10] and [1]. The present study is concerned with detailed geology, petrology and geochemistry of the monzogranite and pegmatites, G. El Fereyid - W. Rahaba, South Eastern Desert, Egypt.

2 Geologic Setting and Petrography

The Precambrian basement rocks exposed in G. El Fereyid - W. Rahaba area area represent a part of the Pan-African belt of the Arabian Nubian Shield (Fig. 3). Gneisses represent the oldest rock unit in the mapped area and crop out in the north eastern and south western parts of the area (Fig. 2) and highly eroded forming low hills. Ophiolitic mélange forms the oldest rock unit in the studied area with gneisses. Ophiolitic mélange of the study area consists of mixture of blocks and masses of varies dimensions of serpentinites and metagabbros. Metavolcanics crop out in the central part of the area as relatively elongate in shape trending roughly in N-S direction moderately to low relief at G. Urga and G. Um Tineidba areas (Fig. 2).



Figure 1. Location map of G. El Fereyid – W. Rahaba area, South Eastern Desert, Egpyt.

Quartz diorite association forms elongated belts of low to moderate relief trending N-S and NNE-SSW at W. Hawdayn and as scattered masses along W. Rahaba (Fig. 2). These rocks are mottled grey to greenish grey, weathered surface; heterogeneous suite with variable grain size and textures. Detailed field mapping of the basement rocks based on the field relations and structure observations, as well as, the petrographical studies revealed that, the area includes older granites (tonalites), younger granites (monzogranite) and pegmatites (Fig. 3). The tonalites are grey in colour, highly exfoliated and forms low hills at the peripheries of the monzogranite. These rocks contain rounded to subrounded xenoliths of diorite and amphibolite up to 30 cm (Fig. 4a). Monzogranite forms the main granitic masses of G. El Fereyid that is elongated in NW- SE directions (Fig. 4a). Monzogranite are leucocratic, medium to coarse-grained, massive, jointed, exfoliated, pinkish grey in color and highly sheared (Fig. 4b). G. El Fereyid monzogranite pluton is bounded from the west and south by W. Rahaba older granites and from the north by G. Marafai shered granites (Fig. 4c). The monzogranite dissected by a series of fault planes especially in the Western side of the pluton, generally trending NW-SE direction and NE-SW and NW-SE trending basic, acidic dykes and quartz veins (Fig. 4d). Pegmatites occur as pockets and veins that crosscut monzogranite (Fig. 4e). They are very coarse grained, reddish in color. Three different types of pegmatite have been defined based on their mineralogy and morphology. The first type of pegmatite is less abundant and found as an irregular body in the southern part of the area (Fig. 3). This muscovite-rich pegmatite is often very coarse grained and contains large individual crystals of its common constituents. It contains the highest radioelement concentrations compared to other rock units as determined from the in situ gamma ray spectrometric survey. The second type of pegmatite is found in the northwestern part of the mapped area, dissecting the monzogranite and it shows a NW – SE strike with variable dips (Fig. 4f). The third type is in the northern part of the mapped area dissecting the tonalites and striking E - W.

Structural analysis shows the presence of a prominent set of folds with axes striking NW to NW and a less prominent set with pronounced ENE and NNW trends. The major and minor faults are mainly thrust, normal and strike slip faults [8]

Tonalite is massive, medium- to coarse-grained, characterized by whitish grey color and porphyritic texture. It consists of plagioclase, K-feldspar and, quartz associated with mafic minerals as hornblende and biotite. Plagioclase is the main feldspar and occurs as anhedral to subhedral tabular crystals of oligoclase (An_{20}) characterized by lamellar and pericline twining, occasionally composed with Carlsbad twining. Most of the crystals are partially saussuritized with variable degrees and some crystals are intensively fractured and fragmented forming mortar texture (Fig. 5a). K-feldspar present as anhedral to subhedral crystals of string perthite (Fig. 5b) partially altered to clay minerals. Quartz occurs as anhedral to subhedral crystals, associating the early-crystallized minerals. Mostly, it shows wavy extinction; some crystals are strained and elongated showing undulose extinction and others are cracked and granulated due to the cataclastic effects (Figs. 5c and 5d). Mafic minerals represented by biotite and hornblende. Biotite occurs as subhedral fractured flakes characterized by pale brown color and faint pleochroism from pale brown to pale greenish brown. It is frequently altered to chlorite and iron oxides along its cleavage planes and includes zircon (Fig. 5c). Muscovite is secondary mica mineral occurring as minute flakes after plagioclase (Fig. 5a) or as very fine flakes of variable size commonly associating the reworked quartz grains (Fig. 5d). Hornblende occurs as subhedarl prismatic crystals, which are strongly pleochroic from pale green; green to yellowish green showing simple twinning (Fig. 5e). The common accessories are zircon, titanite, fluroapatite and opaques (Fig. 5f).



Figure 2. Geologic map of G. El Fereyid - W. Rahaba area, South Eastern Desert, Egypt.



Figure 3. Detailed geologic map of G. El Fereyid - W. Rahaba area, South Eastern Desert, Egypt (Modified after [1]).

Monzogranite are mainly medium to coarse-grained granite with grayish pink color and hypidiomorphic texture. Microscopically, it is composed essentially of potash feldspar, plagioclase, quartz as well as the mafic minerals. Potash feldspars found as microcline, microcline perthite, orthoclase-perthite and perthite. Microcline present as euhedral crystals of medium grain size (3.2 mm length) exhibiting its characteristic cross-hatched twining (Fig. 6a). Microcline petrthite present as subhedral crystals with coarser grain size (5.2 mm length) characterized by cross-hatched twining intergrown with net-like perthitic texture (Fig. 6b). Orthoclase perthite present as euhedral prismatic crystals of medium grain size exhibiting its characteristic Carlsbad twining composed with the perthitic texture (Fig. 6c). Plagioclase (An_{15}) occurs as subhedral to anhedral prismatic crystals of oligoclase showing albite-pericline (Fig. 6d) and albite-carlsbad twinings (Fig.6e). Some crystals are by myrmeckitic textures where quartz excluded as vermicules and others are fractured and partially microclinized or saussiritized enclosing minute crystal of zircon (Fig. 6f). Quartz occurs as subhedral to anhedral cystals with wavy extinction filling the spaces between the early crystallized minerals (feldspars and mafic). Some crystals are elongated and strained showing moderate undulose extinction (Fig. 6g). Mafic minerals are mainly biotite and hornblende. Biotite is more abundant than hornblende; it occurs as subhedral to anhedral flakes pleochroic from X= pale brown to Y=Z= dark brown. Some flakes are foliated along the fractures and associate the elongated quartz crystals (Fig. 6h) and others are partially altered with variable degrees and transformed to chlorite of penninite type with iron oxyhydroxides excluded along the cleavage planes (Fig. 6i). Hornblende is the second mafic mineral in abundance exhibiting its characteristic two-set cleavage in the basal-sectioned crystals. Apatite, epidote, titanite monazite, allanite, and zircon are the main accessory minerals. Titanite occurs as euhedral crystals with sphenoidal form characterized by high relief and high order interference colors (lower fourth order). Monazite occurs as rounded crystal included in biotite and characterized by high relief and high interference colors (upper third order) (Fig. 6j). Zircon occurs as colorless prismatic crystals with high relief; some crystals are surrounded by dark pleochroic haloes within biotite (Fig. 6k).



Figure 4. View showing:- a) Xenoliths of diorite present in tonalities, b): Highly sheared monzogranite (Looking E), c): Intrusive contact between monzogranite and tonalites (Looking NW), d): N-S granitic dyke crosscut the monzogranite (Looking N), e): Pegmatite occurs as vein crosscut the monzogranite (Looking N) and f): Pegmatitic veins crosscut the monzogranite of G. El Fereyid - W. Rahaba area, SED, Egypt. (Looking N).

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Figure 5. Photomicrographs of tonalite, G. El Fereyid - W. Rahaba area, showing:- a) Crystals of plagioclase showing mortar texture and muscovitization (C. N), b) Crystal of string perthite associated with quartz (C.N), c) Fractured flakes of biotite associated with quartz and enclose crystal of zircon (Zr) (C.N), d) Secondary muscovite associated with reworked quartz (C.N), e) Prismatic crystal of hornblende showing simple twining (C.N) and f) Crystal of titanite (Ti) associated with flakes of biotite (C.N.).

Zircon occurs as colorless prismatic crystals with high relief; some crystals are surrounded by dark pleochroic haloes within biotite (Fig. 6k). Other crystals are characterized by zonation and included in plagioclase. Allanite occurs as elongated euhedral crystal with brown color and masked interference color; fractured by effect of the radioelements (Fig. 6l).

Pegmatite occurs as a vein type (granitic and perthitic pegmatites) hosted within the monzogranite. Granitic pegmatite consists mainly of potash feldspar, quartz, plagioclase, and mafic minerals as pegmatitic crystals with grain size more than 2cm. Potash feldspar occurs as euhedral to subhedral mega-crystals represented mainly by microcline (Fig. 7a) associated with microcline perthite, flame perthite, and antiperthite. Plagioclase is a minor constituent occurring as fine crystals included in the megacrysts of potash feldspar (Fig. 7b). It shows pericline- albite twinning and mainly altered to sericite, sometimes, plagioclase in association with quartz crystals shows myrmeckitic texture. Quartz occurs as skeletal crystals digested by the megacrystals of microcline and exhibiting wavy extinction. Biotite occurs as irregular flakes associating plagioclase filling fractures in the potash feldspar (Fig. 7c) or found as euhedral flakes aligned along the fracture in quartz (Fig. 7d). Fergusonite is the main accessory mineral occurring as euhedral prismatic crystals with its pale brown color included in the megacrysts of quartz (Fig. 7e); it is completely isotropic attributed to metamectization due to its radioactive content. Zircon present as fractured crystals coated by iron oxides and associating plagioclase crystals (Fig. 7f). Perthitic pegmatite is composed mainly of megacrysts (more than 2.5cm) of antiperthite (Fig. 8a), microcline (Fig. 8b) and string perthite (Fig. 8c). Perthite crystals are strained characterized by fracturing, undulose extinction and dissected by fractures of secondary silica. Some of the fractures in the microcline megacrysts filled by biotite flakes and iron oxides (see Fig. 8b). Most of the crystals stained by and/or associated with iron oxide, which may be accompanied by amorphous radioactive minerals (Fig. 8c).



Figure 6. Photomicrographs of monzogranite, G. El Fereyid - W. Rahaba area showing:-a) Euhedral crystal of microcline with cross hatching associating biotite (C.N.), b) Unhedral crystal of net microcline pethite associated with quartz (C.N.), c) Subhedral crystal of orthoclase perthite associated with quartz (C.N.), d) Subhedral crystal of plagioclase with albitic and percline twinings (C.N), e) Subhedral crystal of plagioclase with Carlsbad twining (C.N), f) Partially sussiritized zoned plagioclase enclosing zircon (C.N.), g) Unhedral crystal of quartz showing cracking and undulose extinction (C.N.), h) Stretched flakes of biotite due to compression associated with quartz and plagioclase (C.N.), i) Flakes of biotite partially chloritizted to pennite type of chlorite (C.N.), j) Crystal of monazite (Mon) included in flakes of biotite (C.N.), k) Crystal of zircon with zonation included in plagioclases (C.N.) and l) Crystal of allanite (All) with masked interference color associated with quartz (C.N.).



Figure 7. Photomicrographs of perthitic pegnatites, G. El Fereyid - W. Rahaba area showing:- a) Megacryst of antiperthite (C.N), b) Megacryst of microcline enclosing flake of biotite and iton oxides (C.N), c) Secondary iron oxide associated with amorphous radioactive material included in string perthite (C.N).



Figure 8. Photomicrographs of granitic pegmatites, G. El Fereyid - W. Rahaba area showing:-a) Megacrystal of microcline enclosing quartz (C.N), b) Crystals of plagioclase included in megacrystal of microcline (C.N), c) Irregular flake of biotite associating albite included in megacrystal of microcline (C.N.), d) Elongated flakes of biotite along the fracture of quartz (C.N.), e) Subhedral crystal of Fergusonite (Fer) associated with quartz of granitic pegmatites (P.P.L) and f) Crystal of zircon (Zr) included in crystal of plagioclase (C.N.).

3 Geochemistry

Thirty five (25) samples (Twenty from monzogranite and Fifteen from pegmatites veins) were chemically analyzed and were subjected to petrographic studies. Major oxides, trace elements and rare earth elements were analyzed by ICP-MS in Acme Lab Canada. Data of the major oxides, trace elements and rare earth elements are listed in (Table 1). The geochemistry of the studied area is focused mainly on the monzogranite and the pegmatite of the G. El Fereyid area to clarify their chemical classification, chemical affinity, magma type and tectonic setting.

3.1 Geochemical Classification

The classification of the studied monzogranite and pegmatite were chemically confirmed by plotting the analyses on SiO_2 versus alkalis diagrams after [11] (Fig. 9a). The studied samples plot in the field of granite. On the basis of R1-R2 discrimination diagram of De La-Roche [12], the studied granites and pegmatite fall in monzogranite field (Fig. 9b). The Ba content is plotted against its K/Rb ratio for pegmatite samples (Fig.9c). All samples of the El Fereyid area show an increase in Ba content with increasing K/Rb ratio. The increase in Ba content is two orders of magnitude for the El Fereyid pegmatite. Therefore, the Ba content of pegmatitic feldspar is an important indicator of the

fractionation degree of a pegmatite, as already demonstrated for granitic feldspar by Mehnert and Busch [13]. The Li content is plotted against the K/Rb ratio for pegmatite samples (Fig.9d). The majority of the data points fall in the rare-elements class field.

3.2 Alkaline Affinity

Maniar and Piccoli [14] used $(Al_2O_3)/(Na_2O+K_2O+CaO)$ versus $(Al_2O_3)/(Na_2O+K_2O)$ to distinguish between peraluminous, metaluminous and peralkaline rocks (Fig. 9e). The figure reveals that the studied rocks fall in peraluminous field except three sample of pegmatites plot in the line between peralkaline and metaluminous fields. Miyashiro (1975) used total alkalis versus SiO2 (wt %) diagram to distinguish between alkaline and sub-alkaline igneous rocks (Fig. 9f). The figure reveals that the studied samples plot within the subalkaline field. Chappell and White [15] used the bivariate diagram of Na₂O against K_2O to differentiate between S-type and I-type granites (Fig. 10a). The figure clears that the studied rocks plot within I-type granite.

3.3 Tectonic Seeting

Maniar and Piccoli, [14] published some diagrams to discriminate the tectonic setting of the igneous rocks. On the SiO₂ - Al₂O₃ diagram the studied rock samples plot within the post-orogenic granite field (POG) (Fig. 10b).

3.4 Petrogenesis

The normative Qz-Ab-Or values on the ternary diagrams (Figs. 10c&d) are used to define the watervapor pressure and the temperature isotherms at which the studied granites were generated. The watervapor pressures up to 3K-bar after [16] and 5 to 10 K-bars after [17]. It is clear that the studied rock samples fall in a relatively 2-5 K-bar and temperature ranging about 760-840°C indicating that they were probably formed at moderate levels in the crust. By using the K/Rb ratio against Rb diagram which is useful for comparing samples of different sources (Fig. 10e), this diagram suggests that the crystal fractionation was the most important process during magmatic differentiation. A possible magma source of the studied granitic rocks could be inferred from (Fig. 10f) that shows that, the studied rock samples was formed by partial melting of metagray wackes and partial melting of dacite and tonalite that is found in deeper part of the crust of the Arabian Nubian Shield.

3.5 Trace Elements

The concentrations of some trace elements are normalized to a primitive mantle value of Sun and McDonough [18] (Figs. 11a&b). Monzogranite shows enrichment (peaks) of large ion lithophile elements (LILE; Pb, Rb, Ba, Sr) and high field strength elements (HFSE; Y, Zr, Nd,) and depletion (troughs) of K, P and Ti. Pegmatite shows enrichment (peaks) of large ion lithophile elements (LILE; Pb, Rb, Sr) and high field strength elements (HFSE; Y, Zr, Nd,) and depletion (troughs) of K, P and Ti. Pegmatite shows enrichment (peaks) of large ion lithophile elements (LILE; Pb, Rb, Sr) and high field strength elements (HFSE; Y, Zr, Th, U, Nb) and depletion (troughs) of K, P and Ti. The depletion of Ti is ascribed to fractionation of titanomagnetite.

3.6 Rare Earth Elements

Rare earth elements (REEs) analyses of selected samples of monzogranite and pegmatite (Table 1) show total REEs ranging from 154.35 to 240.06 ppm and 41.02 to 244.36 ppm respectively. Σ LREEs in monzogranite range from 136.1 to 223.76 ppm with average 173.79 ppm, while Σ HREEs range from 14.60 to 27.30 ppm with average 20.75 ppm. Based on the REE data and according to Cullers and Graf [19] the monzogranite of the studied area reveal moderate negative Eu anomaly (Eu/Eu* = 0.38 to 0.67) pattern. It has an average Eu/Sm = 0.2, LREEs/HREEs= 8.79. The pegmatite reveals high negative Eu anomaly (Eu/Eu* = 0.12 to 1.09) pattern. It has an average Eu/Sm =0.077, LREEs/HREEs=0.875. The pattern denotes moderate REEs fractionation and shows a narrow range fractionation of HREEs in both monzogranite and pegmatite. The rare earth elements concentration in pegmatite is less than that of monzogranite due to magmatic fractionation and absence of monazite. Chondrite-normalized REEs patterns relative to chondritic values from Boynton, [20], (Figs. 11c&d), of monzogranite and pegmatite are characterized by strong (Σ LREEs) enrichment and low (Σ HREEs) content due to the presence of some accessory minerals like monazite, apatite, zircon.

3.7 REE Tetrad Effect

The term tetrad effect' in geochemistry refers to the subdivision of the 15 lanthanide elements into four groups in a chondrite nor-malized distribution pattern: (1) La – Ce – Pr – Nd, (2) Pm – Sm – Eu – Gd, (3) Tb – Dy – Ho, and (4) Er – Tm – Yb – Lu, and each group forms a smooth convex (M-type) or concave (W-type) pattern (Masuda et al., 1987). The chonderite-normalized REE patterns of the pegmatite show W-type tetrad effect similar to that quoted by Masuda et al. [21] (Fig. 11d). The tetrad effect in the studied monzogranite and pegmatite were calculated according to Irber equations [22]: $Ce^* = Ce_N/La_N^{2/3}xNd_N^{1/3}$, $Pr^* = Pr_N/(La_N^{1/3}xNd_N^{2/3})$, $Tb^* = Tb_N/(Gd_N^{2/3}xHo_N^{1/3})$,

 $Dy^{*} = Dy_{N} / (Gd_{N}^{1/3}xHo_{N}^{2/3}), TE_{3} = (Tb^{*}xDy^{*})^{1/2}, TE_{1} = (Ce^{*}xPr^{*})^{1/2} \text{ and } TE_{1,3} = (TE_{1}xTE_{3})^{1/2}$

When $TE_{1,3}$ values are higher than 1.1, the rock is considered to show the tetrad effect. The studied monzogranite have $TE_{1,3}$ values less than 1.1, so they do not show tetrad effect (Table 1), while the studied pegmatite has an average $TE_{1,3}$ of 1.62, so it show tetrad effect. The fractionation of elements with similar ionic radius and charge is regarded to be sensitive to change in the melt composition during magma differentiation [23-24].

3.8 Geochemistry of Uranium and Thorium

The relation between U ppm and Th ppm may indicate the enrichment or depletion of U because Th is chemically stable. The relation between the U and Th contents is examined by plotting the U contents against Th contents (Fig.12). The uranium and thorium show a notable variations with a somewhat increase within each rock type with increasing SiO2. The U/Th ratio shows irregular variations within each rock type with SiO₂ and Th enrichment. A better positive correlation between U and Th, which may indicate that uranium, is mainly located in thorium rich accessory minerals such as, uranothorite, thorite, zircon, monazite and allanite.

Rock type	Monzogranites												
S. No.	F1	F2	F3	F5	F7	F11	F15	F16	F17	F18	F21	F22	
Major oxides													
SiO_2	72.3	72.72	71.16	70.88	72.78	72.24	70.98	70.35	71.7	71.4	70.79	71.29	
TiO_2	0.52	0.12	0.38	0.4	0.11	0.43	0.44	0.43	0.13	0.63	0.93	037	
Al_2O_3	14.28	14.8	15.14	15.61	14.41	14.42	15.38	14.92	13.9	15.02	14.92	14.38	
$\mathrm{Fe}_2\mathrm{O}_3^*$	2.96	1.7	2.44	2.45	1.55	2.17	2.15	3.17	3.65	3.02	2.35	2.87	
MnO	0.07	0.018	0.031	0.034	0.018	0.061	0.065	0.059	0.057	0.031	0.064	0.019	
MgO	0.93	0.3	0.63	0.66	0.28	0.8	0.75	0.81	0.71	0.55	0.8	0.65	
CaO	1.02	1	1.22	1.3	1.4	0.9	1.16	1.45	1.42	1.24	1.41	1.4	
Na_2O	4.32	4.1	3.82	3.8	4.2	4.33	4.04	4.47	3.99	3.9	4.16	3.77	
K_2O	3.99	4.71	4.5	4.41	4.75	4.08	4.38	3.76	4.56	3.99	4.51	4.66	
P_2O_5	0.27	0.02	0.08	0.11	0.1	0.23	0.21	0.24	0.14	0.03	0.15	0.16	
L.O.I	0.33	0.51	0.4	0.35	0.3	0.34	0.35	0.34	0.36	0.3	0.46	0.32	
Total	100.99	99.99	99.80	100.00	100.00	100.00	99.90	99.99	100.62	100.1	100.54	99.89	
CIPW norm3													
Qz	29.09	28.31	28.39	28.94	27.07	28.97	27.71	26.32	27.33	30.05	25.54	28.01	
Or	23.58	27.83	26.59	26.06	28.07	24.11	25.88	22.22	26.95	23.58	26.65	27.54	
Ab	36.55	34.69	32.32	32.15	35.54	36.64	34.19	37.82	33.76	33	35.2	31.9	
An	3.3	4.83	5.53	5.33	6.29	2.96	4.38	5.63	6.13	596	6.02	5.9	

 Table 1: Chemical composition of monzogranites and pegmatites from G. El Fereyid - W. Rahaba, South Eastern Desert, Egypt.

С	1.65	1.19	1.96	2.16	0.05	2.79	2.39	1.43	0.15	2.1	0.99	0.97
Di/ Wo	0	0	0	0	0	0	0	0	0	0	0	0
Di /en	0	0	0	0	0	0	0	0	0	0	0	0
Ну	2.32	0.75	1.57	1.57	0.64	1.99	1.87	2.02	1.77	1.37	1.99	1.62
Ilm	0.15	0.04	0.07	0.07	0.04	0.13	0.14	0.13	0.12	0.07	0.14	0.04
Hem	2.96	1.7	2.44	2.44	1.55	2.7	2.15	3.170	3.65	3.02	2.35	2.87
Ac	0	0	0	0	0	0	0	0	0	0	0	0
Ар	0	0.05	0.19	0.25	0.23	0.53	0.49	0.56	0.32	0.07	0.35	0.37
Tn	0	0	0	0	0	0	0	0	0.16	0	0	0
Mt	0	0	0	0	0	0	0	0	0	0	0	0
Trace elements	(ppm)	-				-	-	-	-			
U	4.1	5.7	3.7	2.9	4.8	2.2	2.1	2.5	5.1	3.3	2.7	4.9
Th	16.3	29.1	25.6	24.6	29	11.7	9.8	13	28.6	28.7	20.6	22.3
Cr	228	195	220	162	180	202	190	193	261	195	115	260
Y	22.8	30.2	20.8	40.9	33.6	23.7	23.4	21.1	46	44.5	46.1	37.5
Nb	12.67	14.56	12.2	9.9	8.95	9.11	8.7	7.95	13.56	11.89	16.3	11.3
Zr	49.8	40.8	45.5	39.8	41.2	41.5	37.1	46	43.6	48.9	35.2	35.6
Ni	125.6	120.3	90.8	88.9	144.4	124.5	113.5	111.7	114.5	88.9	63.7	92.6
Ba	922	865	980	695	699	861	986	814	789	988	833	955
Pb	41.98	54.6	44.6	70.2	66.9	39.73	60.78	50.91	38.9	50.19	148.24	65.36
Rb	103.6	90.5	120.6	130.5	96.5	85.8	98.5	90.4	95.8	88.3	131.7	111.1
Sr	282	250	280	285	274	278	276	287	266	260	239	275
Zn	381.8	800.5	655.2	579.8	578.9	426.1	847.8	559.8	933.6	908.7	2267.7	789.9
Ti	0.58	0.56	0.57	0.55	0.72	0.47	0.56	0.52	0.44	0.72	0.74	0.54
Cu	543.7	889.6	879.5	2255.6	807.3	708.2	1448.3	962.8	708.6	908	3835.6	908.6
V	43	40	31	38	30	32	33	32	30	32	31	32
Cs	1.7	0.8	0.6	1.3	1.2	0.8	1.2	1.2	1.3	1.4	3.6	1.5
Be	2	2	2	2	3	2	2	2	2	4	2	3
Sn	11.2	14.6	40.2	30.8	51.3	11.6	21.6	15	30.5	16.8	52.6	32.3
Li	35.4	23.3	29.6	33.6	27.9	20	24.1	22.4	26.1	28.9	34.5	25.6
Та	1.1	0.7	0.6	0.7	0.6	0.7	0.7	0.6	0.8	1.6	1.4	0.6
Co	10.2	7.5	5.5	10.1	6.7	8.5	8.5	8.1	8.6	8.5	6.9	7.9
Hf	1.37	1.35	2.3	2.9	1.9	1.21	1.08	1.21	2.55	2.35	1.14	2.06
Ga	21.26	20.12	21.5	20.14	19.87	20.15	20.52	19.87	19.88	20.82	21.03	21.11
Rare earth elem	ients (ppm))										
La	43.60	28.40	47.90	36.20	53.10	29.90	35.00	36.00	46.80	30.40	40.60	50.10
Ce	93.10	62.50	101.50	102.20	92.20	64.65	72.07	75.38	100.30	70.20	83.22	80.70
Pr	11.40	7.50	11.70	10.20	11.20	7.90	8.50	9.30	11.50	11.20	10.20	9.60
Nd	39.10	30.60	44.50	45.50	38.90	28.20	30.40	32.70	38.80	30.50	34.90	29.90
	0.80	6.20	6.70	0.60	0.70	5.60	5.80	5.90	7.20	5.60	7.00	6.80
Eu	1.20	0.90	1.00	1.00	1.20	1.10	1.20	1.10	0.90	1.10	1.00	0.80
Ga	0.50	4.90	0.10	0.70	0.80	4.90	0.80	4.40	0.00	5.20 1.00	0.20	0.10
1 D	1.00	0.80	0.90 5 70	0.70	0.80	0.90	0.90	0.80	0.90	1.00	6.70	0.70
Dy Ua	4.10	0.5U	0.70 1.40	4.00	4.00	4.60	4.20	3.90	0.00	0.0U	0.70	4.50
по Б.	2.70	1.50	1.40	2.00	0.90	0.90	0.90	0.00	1.30	1.30	1.00	2.00
Vh	2.70	4.10	4.00	2.00	2.70	2.40	2.00	2.10	4.70 5.1	4.00	4.00 5.0	2.00
I D	2.60	4.70	4.90	∠.00 0.3	2.70	2.0	2.30	2.00 0.3	0.1	4.00	0.2	2.20 0.3
Tm	0.5	0.7	0.0	0.3	0.4	0.4	0.4	0.3	0.0	0.8	0.0	0.3
Coochemical na	U.4	0.7	0.7	0.4	0.4	0.4	0.4	0.0	0.0	0.1	0.0	0.4

 GR

Zr/Hf	36.4	30.2	19.8	13.7	21.7	34.3	34.4	38.0	17.1	20.8	30.9	17.3
m Zr/Sr	0.2	0.2	0.2	0.1	0.2	0.1	0.1	0.2	0.2	0.2	0.1	0.1
$\rm Rb/Zr$	2.1	2.2	2.7	3.3	2.3	2.1	2.7	2.0	2.2	1.8	3.7	3.1
$\rm Rb/Sr$	0.4	0.4	0.4	0.5	0.4	0.3	0.4	0.3	0.4	0.3	0.6	0.4
$\mathrm{Sr/Rb}$	2.7	2.8	2.3	2.2	2.8	3.2	2.8	3.2	2.8	2.9	1.8	2.5
Y/Nb	1.8	2.1	1.7	4.1	3.8	2.6	2.7	2.7	3.4	3.7	2.8	3.1
Nb/Ta	11.5	20.80	20.30	14.1	14.9	13.0	12.4	13.30	17.0	7.4	11.6	18.80
Yb/Ta	2.5	6.7	8.2	2.9	4.5	3.6	3.3	3.3	6.6	4.5	3.7	3.7
$\mathrm{Ba/Rb}$	8.9	9.6	8.1	5.3	7.2	10.0	10.0	9.0	8.2	11.2	6.3	8.6
Eu/Sm	0.2	0.1	0.1	0.2	0.2	0.2	0.2	0.2	0.1	0.2	0.1	0.1
K/Rb	319.7	432.0	309.8	280.5	408.6	394.7	369.1	345.3	395.1	375.1	284.3	348.2
Rb/Ba	0.1	0.1	0.1	0.2	0.1	0.1	0.1	0.1	0.1	0.1	0.2	0.1
K/Ba	35.9	45.2	38.1	52.7	56.4	39.3	36.9	38.333	48.0	33.5	44.9	40.5
$\Sigma REEs$	213.20	158.80	237.80	217.00	221.00	154.35	170.47	174.98	231.80	172.7	204.22	195.90
Σ LREEs	195.20	136.10	213.30	201.70	203.30	137.35	152.97	160.38	205.50	149.0	176.92	177.90
Σ HREEs	18.0	22.70	24.50	15.30	17.70	17.0	17.50	14.60	26.30	23.70	27.30	18.0
Σ LREEs	10.84	6.0	8.71	13.18	1149	8.08	8.74	10.98	7.81	6.29	6.48	9.88
$/\Sigma$ HREEs												
Eu/Eu*	0.60	0.50	0.48	0.53	0.59	0.64	0.63	0.67	0.43	0.62	0.46	0.38
$(La/Sm)_n$	4.03	2.88	4.50	3.45	4.99	3.36	3.80	3.84	4.09	3.41	3.65	4.63
$(La/Yb)_n$	10.50	4.07	6.59	12.20	13.26	8.06	10.26	12.14	5.95	4.46	5.26	15.35
$(Ce/Yb)_n$	8.60	3.44	5.36	13.22	8.83	6.69	8.11	9.75	4.90	3.95	4.14	9.49
$(Eu/Yb)_n$	1.22	0.54	0.58	1.42	1.26	1.25	1.48	1.56	0.48	0.68	0.55	1.03
$(\mathrm{Tb}/\mathrm{Yb})_{\mathrm{n}}$	1.60	0.8	0.8	1.50	1.3	1.6	1.7	1.8	0.7	1.0	1.0	1.4
TE_1	1.08	1.01	1.03	1.09	0.97	1.07	1.04	1.06	1.09	1.26	1.06	0.99
TE_3	0.95	0.92	0.86	0.86	0.86	1.07	0.94	1.04	0.90	1.0	0.99	0.79
TE	1.02	0.96	0.94	0.97	0.92	1.07	0.99	1.05	0.99	1 13	1.03	0.89

Table 1 (Continued)

Rock type				Pegmatites veins								
S. No.	F23	F24	F25	F26	F30	F33	F35	F36	B2	B3	B5	B9
Major oxides												
SiO_2	71.09	72.2	70.4	72.09	73.6	72.32	72.36	71.81	77.01	73.89	76.18	74.82
TiO_2	0.39	0.55	0.42	0.18	0.11	0.41	0.42	0.28	0.06	0.04	0.19	0.17
Al_2O_3	14.21	14.04	15.1	15.33	14.4	14.39	14.59	15.11	10.6	12.95	11.36	12.36
$\mathrm{Fe_2O_3}^*$	2.88	2.66	2.44	1.8	1.5	2.16	2.07	2.12	1.57	2.33	2.46	1.9
MnO	0.035	0.062	0.023	0.016	0.016	0.052	0.058	0.021	0.02	0.02	0.05	0.04
MgO	0.82	0.91	0.82	0.42	0.28	0.76	0.7	0.55	0.15	0.15	0.45	0.43
CaO	1.27	1.06	1.44	1.05	1.4	1.15	1.27	1.1	0.34	0.22	0.13	0.18
Na_2O	4.05	4.13	4.22	4.00	4.2	3.62	4.57	3.9	3.99	4.77	3.7	4.03
K_2O	4.33	3.87	3.89	4.71	4.75	4.44	3.32	4.66	3.6	3.86	4.43	4.55
P_2O_5	0.06	0.19	0.14	0.04	0.02	0.18	0.24	0.12	0.01	0.01	0.04	0.04
L.O.I	0.36	0.33	0.41	0.36	0.4	0.32	0.4	0.33	1.65	1.49	0.81	1.08
Total	99.50	100.00	99.28	99.99	100.67	99.80	99.99	100.00	99.00	99.73	99.80	99.60
CIPW norm3												
Qz	27.1	30.27	28.39	28.03	27.72	31.18	29.98	28.43	39.37	28.8	36.87	33.05
Or	26.24	22.87	22.99	27.83	28.07	26.24	19.62	27.54	21.27	28.72	26.18	26.89
Ab	34.27	34.95	33.76	33.85	35.54	30.63	38.67	33	33.76	37.57	31.31	34.1
An	5.12	4.02	6.23	4.95	6.41	4.53	4.73	4.72	0.38	1.03	0.38	0.63
С	0.77	1.58	2.04	1.84	0	1.97	1.74	1.92	0	0.01	0.34	0.57

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Di/ Wo	0	0	0	0	0.24	0	0	0	0.03	0	0	0
Di /en	0	0	0	0	0.21	0	0	0	0.81	0	0	0
Hy	2.04	2.27	2.04	1.05	0.67	1.89	1.74	1.37	0	0.37	1.12	1.07
Ilm	0.07	0.13	0.05	0.03	0.03	0.11	0.12	0.04	0.04	0.04	0.11	0.09
Hem	2.88	2.66	2.44	1.8	1.5	2.16	2.07	2.12	1.57	2.33	2.46	1.9
Ac	0	0	0	0	0	0	0	0	0	0	0	0
Ар	0.14	0.44	0.32	0.09	0.05	0.42	0.56	0.28	0.02	0.02	0.09	0.09
Tn	0.41	0	0	0	0.23	0	0	0	0.09	0	0	0
Mt	0	0	0	0	0	0	0	0	0	0	0	0
Trace elements	(ppm)											
U	5.3	4.8	3.8	4.7	5.5	4.5	2.2	5.6	41	16.9	17.3	40.9
Th	13.6	13.2	15.9	20.2	27.4	26.9	20.3	23.3	167.1	100.9	31	90.7
Cr	114	211	222	215	255	112	264	232	11	9	11	7
Y	45.8	33.5	26.5	25.6	38.9	22.3	22.3	45.2	21.3	80.9	180.4	19.8
Nb	15.11	15.46	14.23	13.25	12.62	9.88	8.45	15.21	696.62	180.98	151.41	450.62
Zr	42.9	37.7	40	48.5	47.5	47.1	45.8	38.9	440.8	233.9	152	163.9
Ni	115.6	121.5	87.9	116.2	62.3	119.8	162.1	142.1	5.1	4.6	6.3	5.1
Ba	622	868	897	789	845	980	698	880	84	255	187	100
Pb	105.6	68.17	66.1	68.9	77.2	82.3	45.6	40.6	17.86	36.22	31.39	15.07
Rb	106.5	120.7	120.6	105.5	109.5	100.3	81.7	88.5	442	458	627.7	575
Sr	240	255	285	272	270	244	270	286	25	36	78	22
Zn	544.6	746.2	847.3	398.6	807.9	427.9	524.9	429.6	81.3	122.5	428	122.9
Ti	0.48	0.69	0.63	0.48	0.66	0.67	0.45	0.66	1.2	2.15	2.32	2.13
Cu	847.3	1220.8	1474.6	1252.3	1228.6	1668.5	886	549.5	40.6	9.8	13.9	26.5
V	38	38	31	32	35	34	31	43	7	3	14	1
Cs	0.7	1.4	1.2	0.9	1.6	0.8	1.4	1.6	1.9	1.8	6.2	5.9
Be	3	2	2	2	3	3	2	2	3	7	8	3
Sn	40.5	22.6	23.3	50.1	17.8	15.6	13.7	32.5	24.7	8.6	18.6	80.1
Li	22.2	31	24.6	22.6	23.3	32.6	26.5	24.6	136.6	156.9	205.6	320.2
Та	0.7	1.6	0.8	0.8	0.7	0.6	0.6	0.8	90.4	18.9	13	84.9
Co	9.2	9.4	7.6	6.8	7.8	8.3	8.9	9.8	1.1	0.8	2.6	0.6
Hf	3.1	1.08	1.87	1.06	1.38	2.66	1.32	1.87	20.29	22.29	5.56	10.98
Ga	21.09	20.9	20.66	19.88	19.92	21.03	19.79	21.22	57.24	40.2	42.6	39.3
Rare earth elem	ients (ppm))										
La	32.60	31.30	40.10	37.50	34.50	50.60	52.90	32.20	9.6	8.9	20.9	8.9
Ce	65.30	67.45	88.50	76.68	65.60	89.30	105.96	94.30	43.38	44.62	95.87	35.97
Pr	8.30	8.10	12.90	9.10	8.90	11.90	12.80	7.20	4.1	7.5	10.3	3.8
Nd	25.60	26.70	40.10	30.70	33.60	33.10	43.40	28.10	10.3	20.8	30.7	9.6
Sm	5.40	5.30	6.50	7.20	6.90	6.20	7.40	7.10	2.5	2.4	10.1	1.6
Eu	0.90	0.90	1.20	1.10	1.20	1.10	1.30	1.20	< 0.1	0.3	0.4	< 0.1
Gd	5.20	5.10	6.70	5.70	6.30	4.80	5.50	6.10	1.4	4.6	8.6	3.8
Tb	0.80	1.00	1.00	0.90	1.10	0.90	0.70	0.90	0.4	0.5	2.7	1.3
Dy	5.70	5.90	6.60	6.30	5.70	3.90	3.50	3.40	3.6	4.7	21.5	20.3
Но	1.30	1.30	1.50	1.30	1.50	0.90	0.80	0.80	1.1	1	5.8	0.9
Er	4.20	4.00	4.70	4.00	3.90	2.70	2.40	2.60	5.3	5.9	20.5	5.9
Yb	4.80	4.9	5.1	4.80	4.80	2.5	2.6	2.00	13.5	5.9	27.2	13.6
Lu	0.6	0.7	0.8	0.6	0.7	0.4	0.4	0.3	0.3	0.3	0.3	0.3
Tm	0.7	0.7	0.8	0.7	0.7	0.4	0.4	0.4	1.4	1.4	4.1	1
Geochemical Pa	rameters											
Zr/Hf	13.8	34.9	21.4	45.8	34.4	17.70	34.7	20.8	21.72	10.49	27.34	14.93

m Zr/Sr	0.2	0.1	0.1	0.2	0.2	0.2	0.2	0.1	17.63	6.50	1.95	7.45
$\rm Rb/Zr$	2.5	3.2	3.0	2.2	2.3	2.1	1.8	2.3	1.0	1.96	4.13	3.51
$\rm Rb/Sr$	0.4	0.5	0.4	0.4	0.4	0.4	0.3	0.3	17.68	12.72	8.05	26.14
$\mathrm{Sr/Rb}$	2.3	2.1	2.4	2.6	2.5	2.4	3.3	3.2	0.06	0.08	0.12	0.04
Y/Nb	3.0	2.2	1.9	1.9	3.1	2.3	2.6	3.0	0.03	0.45	1.19	0.04
Nb/Ta	21.6	9.7	17.80	16.60	18.0	21.50	14.10	19.0	7.70	9.6	11.6	5.3
Yb/Ta	6.9	3.1	6.4	6.0	6.9	4.2	4.3	2.5	0.15	0.31	2.09	0.16
$\mathrm{Ba/Rb}$	5.8	7.2	7.4	7.5	7.7	9.8	8.5	9.9	0.19	0.56	0.30	0.17
Eu/Sm	0.2	0.2	0.2	0.2	0.2	0.2	0.2	0.2	0.04	0.13	0.04	0.06
K/Rb	337.5	267.8	267.80	370.60	360.1	367.50	337.30	437.10	67.61	69.96	58.59	65.69
Rb/Ba	0.2	0.1	0.1	0.1	0.1	0.1	0.1	0.1	5.26	1.80	3.36	5.75
K/Ba	57.8	37.0	36.0	49.60	46.70	37.60	39.50	44.0	355.77	125.7	196.66	377.71
$\Sigma REEs$	161.40	163.35	216.50	186.58	175.40	208.70	240.06	186.60	90.55	104.9	244.36	102.66
Σ LREEs	138.10	139.75	189.30	162.28	150.70	190.20	223.76	170.10	61.65	75.95	149.86	53.06
Σ HREEs	23.30	23.60	27.20	24.30	24.70	16.50	16.30	16.50	28.90	29.0	94.50	49.60
Σ LREEs	5.93	5.92	6.96	6.68	6.10	11.65	13.73	10.31	2.13	2.62	1.59	1.07
$/\Sigma$ HREEs												
Eu/Eu^*	0.52	0.53	0.56	0.52	0.56	0.62	0.62	0.56	0.16	0.28	0.13	0.12
$(La/Sm)_n$	380	3.71	3.88	3.28	3.15	5.13	4.50	2.85	2.42	2.33	1.30	3.50
$(La/Yb)_n$	4.58	4.31	5.30	5.27	4.85	13.65	13.72	10.85	0.48	1.02	0.52	0.44
$(Ce/Yb)_n$	3.52	3.56	4.49	4.13	3.54	9.24	10.54	12.20	0.67	1.58	0.74	0.55
$(Eu/Yb)_n$	0.53	0.52	0.67	0.65	0.71	1.25	1.42	1.71	0.02	0.14	0.04	0.02
$(Tb/Yb)_n$	0.7	0.9	0.9	0.8	1.0	1.6	1.2	2.0	0.6	0.4	0.04	0.4
TE_1	1.11	1.11	1.16	1.07	0.97	1.09	1.06	1.19	1.66	1.66	1.53	1.56
TE_3	0.91	1.04	0.89	0.97	0.90	0.99	0.82	0.87	1.07	0.79	1.19	3.07
TE_{13}	1.0	1.08	1.02	1.02	0.94	1.04	0.93	1.02	1.33	1.14	1.35	2.19

Table 1 (Continued)

Pegmatites veins											
S. No.	B11	B13	B15	B18	B20	B23	B25	B27	B30		
Major oxides											
SiO_2	76.5	76.39	74.92	75.72	75.62	75.97	75.83	75.54	73.85		
TiO_2	0.09	0.03	0.09	0.04	0.15	0.02	0.04	0.02	0.09		
Al_2O_3	11.77	11.45	12.75	11.98	12.17	12.34	11.15	12.36	12.98		
$\mathrm{Fe_2O_3}^*$	1.84	1.03	1.82	1.74	1.82	1.9	2.67	2.14	1.85		
MnO	0.1	0.02	0.1	0.01	0.1	0.02	0.12	0.13	0.03		
MgO	0.07	0.05	0.07	0.07	0.03	0.03	0.07	0.02	0.06		
CaO	0.34	0.18	0.32	0.21	0.09	0.06	0.08	0.01	0.18		
Na_2O	3.24	4.46	3.99	4.57	4.26	4.97	4.13	5.11	4.08		
K_2O	4.75	4.01	4.3	3.85	4.66	4.1	4.2	4.22	4.84		
P_2O_5	0.01	0.01	0.04	0.04	0.01	0.01	0.01	0.01	0.04		
L.O.I	1.02	1.37	1.40	1.80	0.89	0.41	1.30	0.44	1.6		
Total	99.73	99.00	99.80	99.63	99.80	99.83	99.60	99.99	99.60		
CIPW norm3											
Qz	38.67	35.12	34.58	33.91	32.81	31.6	35.78	30.36	31.23		
Or	28.07	23.7	25.41	22.75	27.54	24.23	24.82	24.94	28.6		
Ab	27.42	36.57	33.76	38.67	36.05	40.65	33.97	40.08	43.52		
An	1.62	0	1.3	0.8	0.32	0	0	0	0.63		
С	0.7	0	1.05	0	0	0	0	0	0.8		
Di/ Wo	0	0.19	0	0	0	0.01	0	0	0		

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Di /em 0 0.27 0 0.06 0 0.16 0.26 0.21 0.16 0.05 0.05 0.15 Imm 0.17 0.04 0.17 0.02 0.21 0.04 0.06 0.04 0.06 0.04 0.06 0.04 0.06 0.04 0.06 0.04 0.06 0.04 0.05 0.02 0.01										
Hy 0.17 0 0.17 0.01 0.07 0 0.05 0.05 0.15 Ihm 0.17 0.04 0.07 1.77 1.74 1.82 1.47 1.82 1.47 1.83 0.03 1.85 Ac 0 1.03 0 0 0.02 0.0	Di /en	0	0.27	0	0.06	0	0.16	0.26	0	0
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	Ну	0.17	0	0.17	0.15	0.07	0	0.05	0.05	0.15
Hem. 1.79 0.67 1.77 1.74 1.82 1.47 2.18 0.03 1.85 Ac 0 1.03 0 0 0 1.24 0.86 2.78 0 Ap 0.02 0.01 0.03 1.03 0.01 0.02 0.01 0.02 0.01 0.01 0.02 0.01 0.01 0.01 0.01 0.01 0.	Ilm	0.17	0.04	0.17	0.02	0.21	0.04	0.08	0.04	0.06
Ac 0 1.03 0 0 1.24 0.86 2.78 0 Ap 0.02 0.02 0.02 0.02 0.02 0.02 0.09 Tm 0 0.07 0.07 0.01 0.28 0.37 0 The 6.9 44.6 30.9 5.8 33.9 13.3 21.8 28.5 31 Th 6.03 85.8 109.5 22 119.6 34.3 101.5 215.2 110.6 Cr 10 6 8 7 10 6 9 1.7 11 Y 8.9 13.2 26.5 26.9 23.9 20 27.5 20.6 10.0.5 Nb 198.43 53.24 53.8 31.1 4.8 2.8 36.6 5.7 6.1 Ba 239 72 80 55 190 52 50.30 8.89 30.8 St 48 17	Hem	1.79	0.67	1.77	1.74	1.82	1.47	2.18	0.93	1.85
Ap 0.02 0.02 0.02 0.02 0.02 0.02 0.02 0.03 0 0.01 0 0.01 0 0.01 0 0.01 0.02 0.037 0 Trace elements (ppm) 0 0.01 0.28 0.37 0 U 6.03 45.6 30.9 5.8 33.9 13.3 21.8 28.5 31 Th 60.3 85.8 109.5 22 119.6 34.3 101.5 215.2 110.6 Cr 10 6 8 7 10 6 9 17 11 Y 89.9 13.2 20.5 26.9 23.8 20 27.5 20.6 130.5 Tr 13.2 30.9 25.1 28.5 30.6 5.7 6.1 Ba 23.9 72 80 55 190 52 50 7.1 23.5 Pb 259.85 16.1.7 18.35	Ac	0	1.03	0	0	0	1.24	0.86	2.78	0
Tn 0 0.02 0 0.07 0 0 0 0 0 Trace elements (ppm) U 6.9 44.6 30.9 5.8 33.9 13.3 21.8 28.5 31 Th 60.3 85.8 100.5 22 119.6 34.3 101.5 215.2 110.6 Cr 10 6 8 7 10 6 9 17 11 Y 89.9 13.2 26.5 26.9 23.9 20 27.5 20.6 130.5 Nb 198.43 532.45 190.84 180.94 366.79 448.96 610.63 1425.36 532.6 Zr 133.2 300.3 255 190 52 50 71 235 Ba 239 72 80 55 190 52 53 6.1 Ba 239 72 80.3 30.1 83.8 30.8 30.8 <	Ap	0.02	0.02	0.09	0.02	0.02	0.02	0.02	0.02	0.09
Mt 0.07 0 0 0.01 0.28 0.37 0 Trace elements/ppin U 6.9 44.6 30.9 5.8 33.9 13.3 21.8 28.5 31 Th 60.3 85.8 109.5 22 119.6 34.3 101.5 215.2 110.6 Cr 10 6 8 7 10 6 9 17 11 Y 89.9 13.2 26.5 26.9 23.9 20 27.5 20.6 55.5 53.2 Nt 195.4 32.2 30.3 255.1 285.8 365.7 44.8.96 610.63 1475.36 532.6 Rin 239 72 80 55 190 52 50 71 235 Pb 259.85 16.17 18.35 9.44 24.5 11.3 35.2 56.2 Za 185.7 67.6 91.3 100.8 26.3	Tn	0	0.02	0	0.07	0.04	0	0	-0.01	0
Trace elements (ppm) U 6.9 4.4.6 30.9 5.8 33.9 13.3 21.8 28.5 31 Th 60.3 85.8 109.5 22 110.6 6 9 1.7 110 Y 89.9 13.2 26.5 26.9 23.9 20 27.5 20.6 130.5 Nb 198.43 552.46 190.84 180.94 366.79 448.96 610.03 1425.36 526.16 Zr 133.2 33.8 3.1 4.8 2.8 3.6 57.6 61.1 Ba 23.9 72 80 55 190 52.2 56.0 71 235 Pb 259.85 16.17 18.35 93.4 244.5 11.35 35.2 58.68 80.9 Rb 995.2 475 52.2 530.8 50.9 57.12 880.8 830.8 8.09 8.6.9 8.9 8.0 3.0 5.0	Mt	0.07	0	0.07	0	0	0.01	0.28	0.37	0
U 6.9 44.6 30.9 5.8 33.9 13.3 21.8 28.5 31 Th 60.3 85.8 100.5 22 119.6 34.3 101.5 215.2 110.6 Cr 10 6 8 7 10 6 9 17 11 Y 89.9 13.2 26.5 26.9 23.9 20 27.5 20.6 130.5 Nb 198.43 532.45 190.84 180.94 366.79 448.96 610.63 1425.36 532.61 Zr 13.3 3.1 4.8 2.8 3.6 5.7 6.1 Ba 23.9 72 80 55 190 52 50 71 23.5 Zr 475 522.9 530.8 599 571.2 880.1 898.8 30.8 Sr 48 17 2.2 7 3.10 16.3 345.4 77 23.5	Trace elements	(ppm)								
Th 60.3 85.8 109.5 22 119.6 34.3 101.5 215.2 110.6 Cr 10 6 8 7 10 6 9 17 11 Y 89.9 13.2 26.5 26.9 23.9 20 27.5 20.6 130.5 Nb 198.43 552.45 100.84 180.94 366.79 448.96 610.63 1425.36 532.61 Zr 133.2 390.3 255.1 285.8 365.8 291 554.1 428.8 300.8 Mi 2.5 3.2 3.8 3.1 4.8 2.8 3.6 6.9 Rb 295.2 475 522.9 530.8 599 571.2 880.1 898.8 330.8 Sr 48 17 2.2 16 66 18 7 23 52.42 Cu 36.3 39.1 13.3 162 13.5 31.1 6.6	U	6.9	44.6	30.9	5.8	33.9	13.3	21.8	28.5	31
Cr 10 6 9 17 11 Y 89.9 13.2 26.5 26.9 23.9 20 27.5 20.6 130.5 Nb 198.43 532.45 190.84 180.94 366.79 448.96 610.63 1425.36 532.61 Zr 133.2 30.3 255.1 285.8 365.8 291 554.14 428.8 300.8 557. 61.1 Ba 239 72 80 555 190 52 50 71 235 Pb 250.85 16.17 18.35 93.4 244.5 11.35 35.22 536.36 8.69 Rb 995.2 475 52.9 530.8 599 51.2 88.1 170.7 263.5 Tri 2.56 1.25 2.34 1.8 1.88 1.79 2.16 2.35 2.42 3.5 Zn 36.3 39.1 13.3 16.2 13.5 31.1	Th	60.3	85.8	109.5	22	119.6	34.3	101.5	215.2	110.6
Y 89.9 13.2 26.5 26.9 23.9 20 27.5 20.6 130.5 Nb 198.43 532.45 190.84 180.94 366.79 448.96 610.63 1425.36 532.61 Zr 133.2 390.3 255.1 285.8 365.8 201 554.1 428.8 300.8 Ni 2.5 3.2 3.8 3.1 4.8 2.8 3.6 5.7 6.1 Ba 239 72 80 55 100 52 50 71 235 Pb 259.85 16.17 18.35 9.34 244.5 11.63 345.4 170.7 2.3 55 Zn 48 17 2.2 16 66 18 7 2.3 2.42 Cu 36.3 39.1 13.3 16.2 13.5 31.1 6.6 536 30.6 V 2 2 7 3 12 <1	Cr	10	6	8	7	10	6	9	17	11
Nb 198.43 532.45 190.84 180.94 366.79 448.96 610.63 1425.36 532.61 Zr 133.2 390.3 255.1 285.8 365.8 291 554.1 428.8 390.8 Ni 2.5 3.2 3.8 3.1 4.8 2.8 3.6 5.7 6.17 Ba 239 72 80 55 190 52 50.0 71 235 Pb 259.85 16.17 18.35 9.34 244.5 11.35 35.22 536.36 8.69 Rb 995.2 475 522.9 530.8 599 571.2 880.1 898.8 330.8 Sr 48 17 22 16 66 18 7 23 55 Zn 18.37 67.6 91.3 100.8 1.79 1.64 2.42 2.42 7 3 12 <1	Y	89.9	13.2	26.5	26.9	23.9	20	27.5	20.6	130.5
Zr 133.2 390.3 255.1 285.8 365.8 291 554.1 428.8 390.8 Ni 2.5 3.2 3.8 3.1 4.8 2.8 3.6 5.7 6.1 Ba 239 72 80 55 190 52 50 71 235 Pb 259.85 16.17 18.35 9.34 244.5 11.35 35.22 536.6 8.69 Rb 995.2 475 522.9 530.8 599 571.2 880.1 898.8 330.8 Sr 48 1.7 22 16 66 18 7 23 55 Zn 183.7 67.6 91.3 100.8 263.9 116.3 345.4 170.7 263.5 Ti 2.56 1.25 2.34 1.8 1.8 1.2 <11	Nb	198.43	532.45	190.84	180.94	366.79	448.96	610.63	1425.36	532.61
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	Zr	133.2	390.3	255.1	285.8	365.8	291	554.1	428.8	390.8
Ba 239 72 80 55 190 52 50 71 235 Pb 259.85 16.17 18.35 9.34 244.5 11.35 35.22 536.36 8.69 Rb 995.2 475 522.9 530.8 599 571.2 880.1 898.8 330.8 Sr 48 17 22 16 66 18 7 23 55 Ti 2.56 1.25 2.34 1.8 1.88 1.79 2.16 2.35 2.42 Cu 36.3 39.1 13.3 16.2 13.5 31.1 6.6 536 30.6 V 2 2 7 3 12 <1	Ni	2.5	3.2	3.8	3.1	4.8	2.8	3.6	5.7	6.1
Pb 259.85 16.17 18.35 9.34 244.5 11.35 35.22 536.36 8.69 Rb 995.2 475 522.9 530.8 599 571.2 880.1 898.8 330.8 Sr 48 17 22 16 66 18 7 23 55 Zn 183.7 67.6 91.3 100.8 263.9 116.3 345.4 1770.7 263.5 Ti 2.56 1.25 2.34 1.8 1.88 1.79 2.16 6.5 36 30.6 V 2 2 7 3 12 <1	Ba	239	72	80	55	190	52	50	71	235
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$	Pb	259.85	16.17	18.35	9.34	244.5	11.35	35.22	536.36	8.69
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	Rb	995.2	475	522.9	530.8	599	571.2	880.1	898.8	330.8
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	\mathbf{Sr}	48	17	22	16	66	18	7	23	55
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	Zn	183.7	67.6	91.3	100.8	263.9	116.3	345.4	1770.7	263.5
Cu 36.3 39.1 13.3 16.2 13.5 31.1 6.6 536 30.6 V 2 2 7 3 12 <1	Ti	2.56	1.25	2.34	1.8	1.88	1.79	2.16	2.35	2.42
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$	Cu	36.3	39.1	13.3	16.2	13.5	31.1	6.6	536	30.6
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	V	2	2	7	3	12	<1	1	<1	2
Be 5 3 5 4 8 4 3 1 4 Sn 95.1 18.8 32.8 7.9 30.7 23.7 33.8 50.2 87.3 Li 476.3 72.5 309.6 116.5 689.3 102 1619.2 690.2 985.3 Ta 16.4 71 19.9 15.2 63.8 49.1 79.9 171.4 70.9 Co 0.3 0.5 0.5 0.6 0.9 <0.2	Cs	3.5	1.8	3.6	2.9	3.9	2.8	5.3	2.4	3.5
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	Be	5	3	5	4	8	4	3	1	4
Li 476.3 72.5 309.6 116.5 689.3 102 1619.2 690.2 985.3 Ta 16.4 71 19.9 15.2 63.8 49.1 79.9 171.4 70.9 Co 0.3 0.5 0.5 0.6 0.9 <0.2	Sn	95.1	18.8	32.8	7.9	30.7	23.7	33.8	50.2	87.3
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	Li	476.3	72.5	309.6	116.5	689.3	102	1619.2	690.2	985.3
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	Та	16.4	71	19.9	15.2	63.8	49.1	79.9	171.4	70.9
Hf 6.9 18.77 11.9 10.39 5.7 11.76 29.59 27.56 13.2 Ga 38.03 54.31 48.9 36.16 33.9 69.34 59.3 65.45 40.9 Rare earth elements (ppm)La 18.8 9.3 6.8 5.1 17.3 2.6 5.7 2.4 6.9 Ce 48.88 39.35 60.5 23.85 30.88 14.22 28.97 12.13 75.19 Pr 6.1 4.6 5.9 2.8 6.1 1.7 3.7 1.5 8.9 Nd 20.8 10.3 10.3 6.3 29.9 3.3 8.2 3.1 25.39 Sm 5.4 2.6 1.1 1.7 0.9 1.1 2.3 0.9 5.9 Eu 0.3 <0.1 0.3 <0.1 0.4 <0.1 <0.1 <0.1 0.4 Gd 4.6 1.1 1.7 0.9 1.1 2.3 0.9 5.9 Eu 0.3 <0.1 0.3 <0.1 0.4 <0.1 <0.1 <0.1 0.4 Gd 4.6 1.1 5.6 1.1 1.4 0.6 1.1 0.7 6.8 Tb 1.3 0.5 2.1 0.4 0.2 0.4 0.5 0.3 1.5 Dy 10.2 3 10.6 3.6 9.8 3 4.7 3.5 11.6 Ho 2.5 0.9 $0.$	Co	0.3	0.5	0.5	0.6	0.9	< 0.2	0.4	0.5	1.1
Ga 38.03 54.31 48.9 36.16 33.9 69.34 59.3 65.45 40.9 Rare earth elements (ppm) La 18.8 9.3 6.8 5.1 17.3 2.6 5.7 2.4 6.9 Ce 48.88 39.35 60.5 23.85 30.88 14.22 28.97 12.13 75.19 Pr 6.1 4.6 5.9 2.8 6.1 1.7 3.7 1.5 8.9 Nd 20.8 10.3 10.3 6.3 29.9 3.3 8.2 3.1 25.39 Sm 5.4 2.6 1.1 1.7 0.9 1.1 2.3 0.9 5.9 Eu 0.3 <0.1	Hf	6.9	18.77	11.9	10.39	5.7	11.76	29.59	27.56	13.2
Rare earth elements (ppm)La18.89.36.85.117.32.65.72.46.9Ce48.8839.3560.523.8530.8814.2228.9712.1375.19Pr6.14.65.92.86.11.73.71.58.9Nd20.810.310.36.329.93.38.23.125.39Sm5.42.61.11.70.91.12.30.95.9Eu0.3<0.10.3<0.10.4<0.1<0.1<0.10.4Gd4.61.15.61.11.40.61.10.76.8Tb1.30.52.10.40.20.40.50.31.5Dy10.2310.63.69.834.73.511.6Ho2.50.90.912.911.41.11.4Er9.84.111.33.912.43.95.55.919.5Yb20.19.214.96.913.47.113.415.627.2Lu0.30.30.30.30.30.30.30.30.30.3Tm2.811.4110.91.61.72.8Geochemical Parameters27.7Hf19.3020.7921.4427.5164.1824.7418.73 <t< td=""><td>Ga</td><td>38.03</td><td>54.31</td><td>48.9</td><td>36.16</td><td>33.9</td><td>69.34</td><td>59.3</td><td>65.45</td><td>40.9</td></t<>	Ga	38.03	54.31	48.9	36.16	33.9	69.34	59.3	65.45	40.9
La18.89.36.85.117.32.65.72.46.9Ce48.8839.3560.523.8530.8814.2228.9712.1375.19Pr6.14.65.92.86.11.73.71.58.9Nd20.810.310.36.329.93.38.23.125.39Sm5.42.61.11.70.91.12.30.95.9Eu0.3<0.1	Rare earth elem	nents (ppm))				-	-	-	
Ce48.8839.3560.523.8530.8814.2228.9712.1375.19Pr6.14.65.92.86.11.73.71.58.9Nd20.810.310.36.329.93.38.23.125.39Sm5.42.61.11.70.91.12.30.95.9Eu0.3<0.1	La	18.8	9.3	6.8	5.1	17.3	2.6	5.7	2.4	6.9
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	Ce	48.88	39.35	60.5	23.85	30.88	14.22	28.97	12.13	75.19
Nd20.810.310.36.329.93.38.23.125.39Sm 5.4 2.6 1.1 1.7 0.9 1.1 2.3 0.9 5.9 Eu 0.3 <0.1 0.3 <0.1 0.4 <0.1 <0.1 <0.1 0.4 Gd 4.6 1.1 5.6 1.1 1.4 0.6 1.1 0.7 6.8 Tb 1.3 0.5 2.1 0.4 0.2 0.4 0.5 0.3 1.5 Dy 10.2 3 10.6 3.6 9.8 3 4.7 3.5 11.6 Ho 2.5 0.9 0.9 1 2.9 1 1.4 1.1 1.4 Er 9.8 4.1 11.3 3.9 12.4 3.9 5.5 5.9 19.5 Yb 20.1 9.2 14.9 6.9 13.4 7.1 13.4 15.6 27.2 Lu 0.3 0.3 0.3 0.3 0.3 0.3 0.3 0.3 0.3 Tm 2.8 1 1.4 1 1 0.9 1.6 1.7 2.8 Geochemical Parameters Zr/Hf 19.30 20.79 21.44 27.51 64.18 24.74 18.73 15.56 29.61 Zr/Sr 2.78 22.96 11.60 17.86 5.54 16.17 79.16 18.64 7.11	Pr	6.1	4.6	5.9	2.8	6.1	1.7	3.7	1.5	8.9
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$	Nd	20.8	10.3	10.3	6.3	29.9	3.3	8.2	3.1	25.39
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	Sm	5.4	2.6	1.1	1.7	0.9	1.1	2.3	0.9	5.9
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Eu	0.3	< 0.1	0.3	< 0.1	0.4	< 0.1	< 0.1	< 0.1	0.4
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Gd	4.6	1.1	5.6	1.1	1.4	0.6	1.1	0.7	6.8
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	Tb	1.3	0.5	2.1	0.4	0.2	0.4	0.5	0.3	1.5
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Dy	10.2	3	10.6	3.6	9.8	3	4.7	3.5	11.6
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	Но	2.5	0.9	0.9	1	2.9	1	1.4	1.1	1.4
Yb 20.1 9.2 14.9 6.9 13.4 7.1 13.4 15.6 27.2 Lu 0.3 </td <td>Er</td> <td>9.8</td> <td>4.1</td> <td>11.3</td> <td>3.9</td> <td>12.4</td> <td>3.9</td> <td>5.5</td> <td>5.9</td> <td>19.5</td>	Er	9.8	4.1	11.3	3.9	12.4	3.9	5.5	5.9	19.5
Lu 0.3	Yb	20.1	9.2	14.9	6.9	13.4	7.1	13.4	15.6	27.2
Tm 2.8 1 1.4 1 1 0.9 1.6 1.7 2.8 Geochemical Parameters Zr/Hf 19.30 20.79 21.44 27.51 64.18 24.74 18.73 15.56 29.61 Zr/Sr 2.78 22.96 11.60 17.86 5.54 16.17 79.16 18.64 7.11	Lu	0.3	0.3	0.3	0.3	0.3	0.3	0.3	0.3	0.3
Geochemical Parameters Zr/Hf 19.30 20.79 21.44 27.51 64.18 24.74 18.73 15.56 29.61 Zr/Sr 2.78 22.96 11.60 17.86 5.54 16.17 79.16 18.64 7.11	Tm	2.8	1	1.4	1	1	0.9	1.6	1.7	2.8
Zr/Hf 19.30 20.79 21.44 27.51 64.18 24.74 18.73 15.56 29.61 Zr/Sr 2.78 22.96 11.60 17.86 5.54 16.17 79.16 18.64 7.11	Geochemical Pa	arameters					•	•	•	
Zr/Sr 2.78 22.96 11.60 17.86 5.54 16.17 79.16 18.64 7.11	Zr/Hf	19.30	20.79	21.44	27.51	64.18	24.74	18.73	15.56	29.61
	Zr/Sr	2.78	22.96	11.60	17.86	5.54	16.17	79.16	18.64	7.11

Rb/Zr	7.47	1.22	2.05	1.86	1.64	1.96	1.59	2.10	0.85
$\rm Rb/Sr$	20.73	27.94	23.77	33.18	9.08	31.73	125.73	39.08	6.01
$\rm Sr/Rb$	0.05	0.04	0.04	0.03	0.11	0.03	0.01	0.03	0.17
Y/Nb	0.45	0.02	0.14	0.15	0.07	0.04	0.05	0.01	0.25
Nb/Ta	12.10	7.5	9.6	11.90	5.8	9.1	7.6	8.30	7.5
Yb/Ta	1.23	0.13	0.75	0.45	0.21	0.14	0.17	0.09	0.38
$\mathrm{Ba/Rb}$	0.24	0.15	0.15	0.10	0.32	0.09	0.06	0.08	0.71
Eu/Sm	0.06	0.04	0.27	0.06	0.44	0.09	0.04	0.11	0.07
$\rm K/Rb$	39.62	70.08	68.26	60.21	64.58	59.59	39.62	38.98	121.46
$\rm Rb/Ba$	4.16	6.60	6.54	9.65	3.15	10.98	17.60	12.66	1.41
K/Ba	164.98	462.34	446.19	581.09	203.60	654.56	697.31	493.40	170.97
$\Sigma REEs$	79.57	87.65	135.0	58.85	130.18	41.02	79.57	51.53	193.18
Σ LREEs	100.28	66.25	48.97	39.85	85.48	23.02	48.97	20.13	122.68
Σ HREEs	54.90	21.40	30.60	19.0	44.70	18.0	30.60	31.40	70.50
Σ LREEs	1.83	3.10	1.60	2.10	1.91	1.28	1.60	0.64	1.74
$/\Sigma$ HREEs									
Eu/Eu*	0.18	0.18	0.37	0.22	1.09	0.38	0.19	0.39	0.19
$(\rm La/Sm)_n$	2.19	2.25	3.89	1.89	12.09	1.49	1.56	1.68	0.74
$(La/Yb)_n$	0.63	0.68	0.31	0.50	0.87	0.25	0.29	0.1	0.17
$(Ce/Yb)_n$	0.63	1.11	1.05	0.89	0.60	0.52	0.56	0.2	0.72
$(Eu/Yb)_n$	0.04	0.03	0.06	0.04	0.08	0.04	0.02	0.02	0.04
$(Tb/Yb)_n$	0.3	0.2	0.6	0.3	0.1	0.2	0.2	0.1	0.2
TE_1	1.20	1.89	3.10	1.98	0.83	2.31	2.08	2.15	2.68
TE_3	1.19	1.36	2.32	1.26	0.77	1.56	1.36	1.29	1.49
$TE_{1,2}$	1.19	1.60	2.68	1.58	0.80	1.90	1.68	1.66	2.0

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Figure 9a): Na2O+K2O vs. SiO2 variation diagram [11]. Symbols are the same for all next diagrams (○ monzogranite and ◇ pegmatites of G. El Fereyid – W. Rahaba area), b): R1-R2 diagram after [12], c): Ba vs K/Rb diagram according to Mehnert and Busch [13], d): Li vs K/Rb diagram according to Cerny and Burt [25], e): Molar Al2O3/(CaO+Na2O+K2O) versus Al2O3/(Na2O+K2O) [14] and f): Alkali-SiO2 variation diagram [26], G. El Fereyid - W. Rahaba, South Eastern Desert, Egypt.



Figure 10. a): Na2O-K2O discrimination diagram after [15], b): SiO2 vs Al2O3 diagram after [14], c): Normative Qz-Ab-Or ternary giagram. The dashed line represents the variation in position of the minimum melting points in the granite system at different water vapor pressures. [16] and [17], d): Normative Qz-Ab-Or isobaric equilibrium ternary diagram after [16], e): K/Rb versus Rb binary diagram of the studied samples and f): K2O - SiO2 diagram. Field of partial melts of different rock sources are after [27], G. El Fereyid - W. Rahaba, South Eastern Desert, Egypt.



Figure 11. a): Normalized multi-elements patterns of the monzogranite [18], b): Normalized multi-elements patterns of the pegmatites [18], c): Rare earth element distribution pattern of monzogranite. Normalized to chondrite of [20] and d): Rare earth element distribution pattern of pegmatites. Normalized to chondrite of [20], G. El Fereyid - W. Rahaba, South Eastern Desert, Egypt.

4 Conclusions

The basement rocks exposed in the area include tonalities, monzogranite and pegmatites. Tonalites are whitish grey to grey, medium to coarse-grained, variable strained, highly weathered and exfoliated and porphyritic texture. It consists of plagioclase, K-feldspar and, quartz associated with mafic minerals as hornblende and biotite. Monzogranite forms the main granitic masses of G. El Fereyid - W. Rahaba that is elongated in NW- SE directions. Monzogranite is leucocratic, medium to coarse-grained, massive, jointed, exfoliated, pinkish grey in color and highly sheared and composed essentially of potash feldspar, plagioclase, quartz as well as the mafic minerals. Pegmatite can be considered as the most abundant rock units bearing anomalous radioactive signature. The studied pegmatitic rocks occur as a vein type (granitic and perthitic pegmatites) hosted within the monzogranite. The general trend of the study younger granites suggests that pegmatite rock unit is the most differentiated in the study area and the studied younger granites of G. El Fereyid are low calcium, iron oxides and MgO contents but they are rich in alkalies (Potassium and sodium oxides). Monzogranite shows enrichment (peaks) of large ion lithophile elements (LILE; Pb, Rb, Ba, Sr) and high field strength elements (HFSE; Y, Zr, Nd,) and depletion (troughs) of K, P and Ti. Pegmatite shows enrichment (peaks) of large ion lithophile elements (LILE; Pb, Rb, Sr) and high field strength elements (HFSE; Y, Zr, Th, U, Nb) and depletion (troughs) of K, P and Ti.



Figure 12. Correlation between U, Th and U/Th ratio for monzogranite and pegmatites, G. El Fereyid - W. Rahaba, South Eastern Desert, Egypt.

They exhibit high K-calc alkaline magma and peraluminous characteristics and also considered as post orogenic granites (POG) and formed as a result of partial melting of the lower crust that was mainly amphibolites. The estimated temperature, using both the apatite and zircon models, is 760–840 C. This temperature is consistent with partial melting reactions of amphibole-bearing rock at depths of >30 km. They were subjected to many alteration processes including silicification, kaolinization and hematitization. The monzogranite reveal small to moderate negative Eu anomaly, while the pegmatite reveal moderate to large negative Eu anomaly. Hydrothermal alteration during post-magmatic stages is also identified through the development of a W-type tetrad effect in the REE pattern of pegmatite. The distinct behavior of the most popular isovalent elements ratio (Zr/Hf, Nb/Ta, Th/U, Y/Ho) during hydrothermal alteration seems to be non-chondritic and is no longer exclusive the control of ionic radius and charge.

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